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Key indicators

Single-crystal X-ray study $T=293~{\rm K}$ Mean $\sigma({\rm C-C})=0.002~{\rm \AA}$ R factor = 0.045 wR factor = 0.113 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://iournals.jucr.org/e.

N-[(2-Hydroxynaphthalen-1-yl)(phenyl)methyl]-acetamide

The crystal structure of the title compound, $C_{19}H_{17}N_1O_2$, obtained *via* a one-pot synthesis, is stabilized by intramolecular $N-H\cdots O$ and intermolecular $O-H\cdots O$ hydrogen bonds.

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Comment

Multi-component reactions (MCRs) have attracted considerable attention in terms of the saving of both energy and raw materials (Devi & Bhuyan, 2004). They have merits over multi-step reactions in several aspects, including the simplicity of a one-pot procedure, possible structural variations and in building up complex molecules (Domling & Ugi, 2000). Here we report the synthesis and crystal structure of the title compound (I) (Fig. 1), obtained by a three-component condensation reaction of 2-naphthol, benzaldehyde and acetamide under solvent-free conditions.

In the molecule of the title compound, (I) (Fig. 1), the bond distances and angles are within their normal ranges (Allen *et al.* 1987, Bazgir *et al.* 2006). The dihedral angles between the rings A (C14–C19), B (C4–C8/C13) and C (C8–C13) are A/B = 82.10 (2), A/C = 80.08 (3) and B/C = 2.25 (4)°. An intramolecular N—H···O interaction and an intermolecular O—H···O hydrogen bond (Table 1) help to establish the crystal packing (Fig. 2).

Experimental

2-Naphthol (1 mmol), benzaldehyde (1 mmol), acetamide (1.4 mmol) and p-toluenesulfonic acid (0.1 g) were mixed and the reaction mixture was placed in a screw-capped vial and heated at 373 K for 3 h. After cooling, the reaction mixture was washed with water and then recrystallized from EtOAc/hexane (1:2) to afford the pure product in 73% yield.

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Crystal data

$C_{19}H_{17}NO_2$	Z = 4
$M_r = 291.34$	$D_x = 1.291 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
a = 11.835 (3) Å	$\mu = 0.08 \text{ mm}^{-1}$
b = 7.3428 (10) Å	T = 293 (2) K
c = 18.034 (3) Å	Block, colorless
$\beta = 106.958 \ (16)^{\circ}$	$0.50 \times 0.35 \times 0.25 \text{ mm}$
$V = 1499.1 (5) \text{ Å}^3$	

Data collection

Stoe IPDS-II diffractometer	9203 measured reflections
ω scans	3534 independent reflections
Absorption correction: numerical	2932 reflections with $I > 2\sigma(I)$
(X-RED32 and X-SHAPE; Stoe)	$R_{\rm int} = 0.023$
& Cie, 2005)	$\theta_{\rm max} = 27.9^{\circ}$
$T_{\min} = 0.960, T_{\max} = 0.980$	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0468P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.045$	+ 0.3551P
$wR(F^2) = 0.114$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.06	$(\Delta/\sigma)_{\text{max}} = 0.003$
3534 reflections	$\Delta \rho_{\text{max}} = 0.19 \text{ e Å}^{-3}$
267 parameters	$\Delta \rho_{\min} = -0.16 \text{ e Å}^{-3}$
All H-atom parameters refined	

Table 1 Hydrogen-bond geometry (Å, °).

D $ H$ $\cdot \cdot \cdot A$	<i>D</i> —Н	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	D $ H$ $\cdot \cdot \cdot A$
$N1-H1\cdots O2$ $O2-H2\cdots O1^{i}$	0.889 (15)	2.118 (15)	2.7052 (16)	122.8 (13)
	0.87 (2)	1.82 (2)	2.6807 (16)	175 (3)

Symmetry code: (i) x, y + 1, z.

All of the H atoms were located in a difference sythesis and their positions and $U_{\rm iso}$ values were freely refined.

Data collection: *X-AREA* (Stoe & Cie, 2005); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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References

Allen, F. H., Kennard, O., Watson, D. G., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans.* 2, pp. S1–19.

Bazgir, A., Amani, V. & Khavasi, H. R. (2006). *Acta Cryst.* E**62**, o3533–o3534. Devi, I. & Bhuyan, P. J. (2004). *Tetrahedron Lett.* **45**, 8625–8627.

Domling, A. & Ugi, I. (2000). Angew. Chem. Int. Ed. 39, 3168–3210.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837–838.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

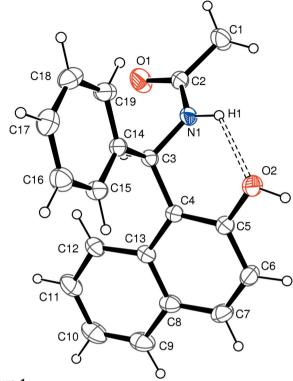
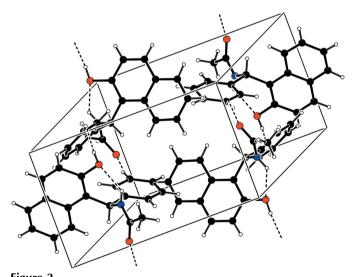


Figure 1
The molecular structure of (I) with displacement ellipsoids drawn at the 30% probability level (arbitrary spheres for the H atoms). The double dashed line indicates a hydrogen bond.



A packing diagram of (I). Hydrogen bonds are shown as dashed lines.

Stoe & Cie (2005). X-AREA (Version 1.31), X-RED32 (Version 1.28b) and X-SHAPE (Version 2.05). Stoe & Cie, Darmstadt, Germany.